

New Bulk Glassy Ni-Based Alloys with High Strength of 3000 MPa

Tao Zhang and Akihisa Inoue

Institute for Materials Research, Tohoku University, Sendai 980-8577, Japan

New Ni-based bulk glassy alloys with high strength and good ductility were synthesized for the first time in Ni–Nb–Ti–Zr base system by the mold-clamp or copper mold casting method. The bulk glassy $\text{Ni}_{53}\text{Nb}_{20}\text{Ti}_{10}\text{Zr}_8\text{Co}_6\text{Cu}_3$ alloy has a rod shape with diameters up to 3 mm or a sheet shape with thickness up to 1 mm. The glass transition temperature (T_g) and the supercooled liquid region defined by the difference between T_g and crystallization temperature (T_x), $\Delta T_x (= T_x - T_g)$ are 846 and 51 K, respectively, and no distinct change in T_g , T_x and ΔT_x with sample diameter is seen. The Ni-based alloy is located in the vicinity of eutectic composition and has a high reduced glass transition temperature (T_g/T_m) of 0.67. The Ni-based bulk glassy alloy also exhibits good mechanical properties, *i.e.*, tensile fracture strength of 2700 MPa, tensile fracture elongation of 2.1%, compressive fracture strength of 3010 MPa and compressive fracture elongation of 2.4%. It is noticed that the tensile fracture strength is the highest among all bulk glassy alloys developed up to date. The success of synthesizing the new Ni-based bulk glassy alloy with good mechanical properties is promising for future uses as a new type of high strength material.

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1. Introduction

Since the success of forming bulk glassy alloys with a large supercooled liquid region before crystallization and good mechanical properties in lanthanide (Ln)-¹⁾ and Mg-²⁾ based systems by a copper mold casting method in late 1980's, great attention has been devoted to synthesis and properties of a new bulk glassy alloy. It is well known that a number of bulk glassy alloys were subsequently developed in the order of Zr-,^{3,4)} Ti-,⁵⁾ Hf-,⁶⁾ Fe-,⁷⁾ Pd–Cu-,⁸⁾ Pd–Fe-,⁹⁾ Co-,¹⁰⁾ Ni-¹¹⁾ and Cu-¹²⁾ based alloy systems and the maximum sample thickness reaches as large as about 10 mm¹³⁾ for the Ln-based alloys, 12 mm¹⁴⁾ for the Mg-based alloys, 15 to 30 mm^{4,15)} for the Zr-based alloys, 5 mm¹⁶⁾ for the Fe-based alloys, 75 mm¹⁷⁾ for the Pd–Cu-based alloys, 2 mm for the Co-¹⁸⁾ and Ni-¹⁹⁾ based alloys and 5 mm²⁰⁾ for the Cu-based alloys. Among these bulk glassy alloys, it has been reported that high tensile strength combined with good ductility is obtained for the Ln-, Zr-, Hf-, Pd–Cu- and Cu-based alloys and good soft magnetic properties are obtained for the Fe- and Co-based alloys.^{21–25)} When we pay attention to high-strength glassy alloys with good ductility, the tensile strength level has been reported to be 1200 MPa for the Ln-based alloys,²⁴⁾ 1500 to 1700 MPa for the Zr-, Hf- and Pd–Cu-based alloys²⁴⁾ and 2000 to 2500 MPa^{12,20)} for the Cu-based alloys. Although much higher mechanical strength is expected to be obtained for the ferrous group glassy alloys in Fe-, Co- and Ni-based alloy systems, high tensile strength exceeding 2500 MPa has not been achieved because of their poor ductility leading to fracture within elastic limit. The expectation of achieving high mechanical strength in the ferrous alloy group by improving ductility is supported from some experimental data that high compressive fracture strength of about 2500 MPa has been obtained for Ni-based metal-metalloid type bulk glassy alloys.¹⁹⁾ Based on the recent data on the new alloy compositions of the Cu-based bulk glassy alloys with high tensile strength of 2000 to 2500 MPa,^{12,20,25)} we have subsequently performed a study to synthesize a new bulk glassy

alloy with much higher tensile strength in the ferrous alloy groups by finding a new alloy composition at which a good ductility without fracture in the elastic limit is obtained. We have succeeded for the first time in synthesizing new bulk glassy Ni-based alloys with high tensile strength of 2700 MPa and high compressive fracture strength of 3010 MPa. This paper intends to present the compositions, thermal stability and mechanical properties of the high-strength Ni-based bulk glassy alloys.

2. Experimental Procedure

Multicomponent alloy ingots with composition of $\text{Ni}_{53}\text{Nb}_{20}\text{Ti}_{10}\text{Zr}_8\text{Co}_6\text{Cu}_3$ were prepared by arc melting the mixtures of pure metals in an argon atmosphere. The composition represent nominal atomic percentages. Bulk glassy alloys were prepared in a cylindrical rod shape with diameters up to 5 mm by the copper mold casting method and in a sheet shape with thickness up to 3 mm by the mold-clamp casting method.²⁶⁾ The glassy alloy ribbon was also prepared by melt spinning. The glassy phase was identified by X-ray diffraction. The absence of microscale crystalline phase was examined by optical microscopy. The thermal stability associated with glass transition, supercooled liquid region and crystallization was examined by differential scanning calorimetry (DSC) at a heating rate of 0.67 K/s. The melting temperature (T_m) was measured by differential thermal analysis (DTA). Mechanical properties were measured with an Instron testing machine at a strain rate of $8.3 \times 10^{-4} \text{ s}^{-1}$ for tensile test and $4.8 \times 10^{-4} \text{ s}^{-1}$ for compressive test. Fracture surface was examined by scanning electron microscopy (SEM).

3. Results and Discussion

Figure 1 shows the shape and outer appearance of cast bulk glassy $\text{Ni}_{53}\text{Nb}_{20}\text{Ti}_{10}\text{Zr}_8\text{Co}_6\text{Cu}_3$ alloys in a rod shape of 3 mm in diameter and in a sheet shape of 1 mm in thickness. The outer surface is smooth and neither appreciable rugged-

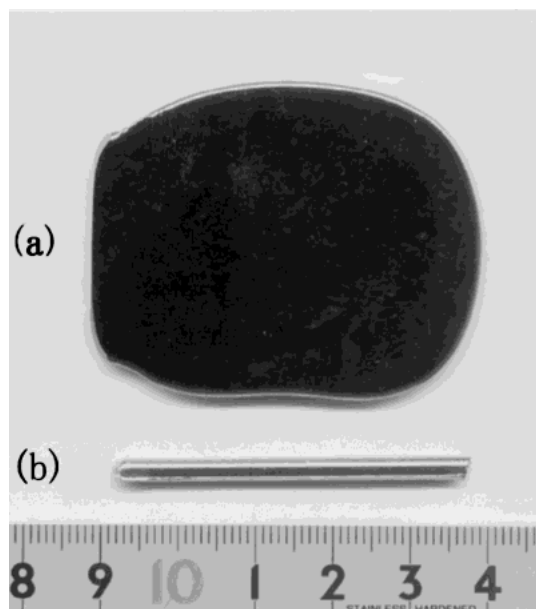


Fig. 1 Outer shape and surface morphology of cast $\text{Ni}_{53}\text{Nb}_{20}\text{Ti}_{10}\text{Zr}_8\text{Co}_6\text{Cu}_3$ alloys in rod and sheet shapes. (a) ϕ 3 mm rod and (b) 1 mm thickness sheet.

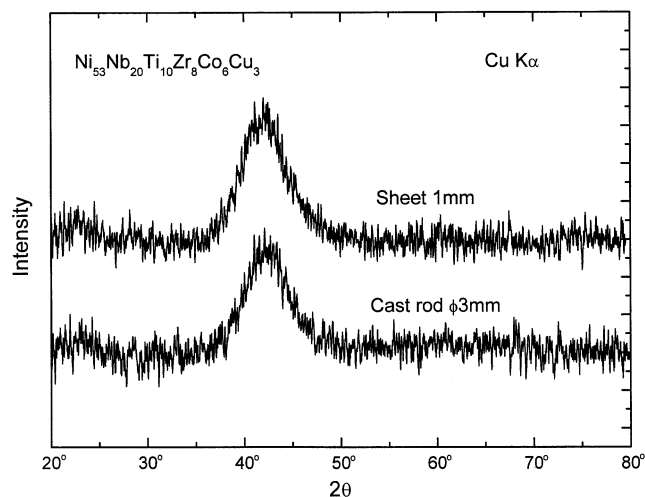


Fig. 2 X-ray diffraction patterns of the cast $\text{Ni}_{53}\text{Nb}_{20}\text{Ti}_{10}\text{Zr}_8\text{Co}_6\text{Cu}_3$ alloys in rod and sheet shapes.

ness nor concave is seen, indicating the formation of a glassy phase. The X-ray diffraction patterns of the bulk glassy alloys are shown in Fig. 2. The diffraction patterns consist only of broad peaks and no appreciable sharp diffraction peak due to a crystalline phase is seen. The feature of the X-ray diffraction patterns is independent of rod diameter. Figure 3 shows DSC curves of the bulk glassy alloys, together with the data of the corresponding melt-spun glassy alloy ribbon. All the samples exhibit a sequent phase transition of an endothermic reaction due to glass transition, followed by a large supercooled liquid region and then an exothermic reaction due to crystallization. The glass transition temperature (T_g) and onset temperature of crystallization (T_x) are 846 and 897 K, respectively for the rod sample with a diameter of 3 mm. No appreciable change in T_g and T_x was recognized in the present diameter range. In addition, one can see that the feature of the DSC curves as well as the T_g and T_x values is the same as those

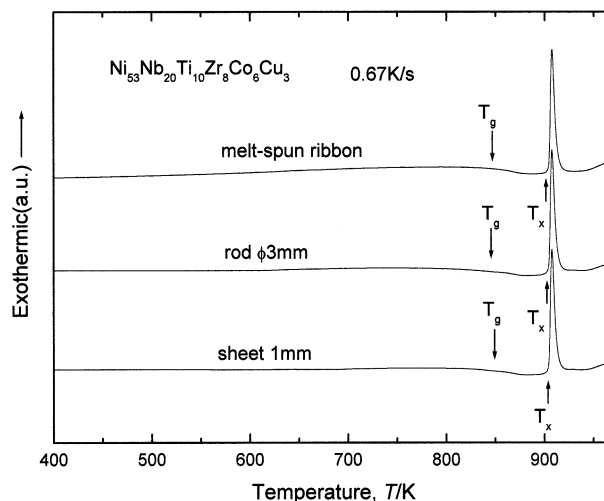


Fig. 3 DSC curves of the cast bulk glassy $\text{Ni}_{53}\text{Nb}_{20}\text{Ti}_{10}\text{Zr}_8\text{Co}_6\text{Cu}_3$ alloys in rod and sheet shapes.

for the melt-spun glassy alloy ribbon. Here, it is noticed that the bulk glassy $\text{Ni}_{53}\text{Nb}_{20}\text{Ti}_{10}\text{Zr}_8\text{Co}_6\text{Cu}_3$ alloys exhibit a large supercooled liquid region of 51 K which is defined by the difference between T_g and T_x , $\Delta T_x (= T_x - T_g)$. The ΔT_x value is nearly the same as the largest value for a number of Ni-based metal-metalloid type glassy alloys such as Ni–Cr–Mo–P–B,²⁷⁾ Ni–Nb–Cr–Mo–P–B²⁷⁾ and Ni–Zr–Ti–Sn–Si²⁸⁾ systems. Although the ΔT_x value is considerably smaller than the largest values for the Zr–Al–Ni–Cu²⁹⁾ and Pd–Cu–Ni–P⁸⁾ glassy alloys, the relatively large ΔT_x value indicates that the supercooled liquid of the new metal-metal type Ni-based alloy has a rather high stability against crystallization.

We further examined the reduced glass transition temperature (T_g/T_m) defined by the ratio of glass transition temperature (T_g) to melting temperature (T_m) by measuring the melting temperature with a differential thermal analyzer (DTA). As exemplified for the DTA curve in Fig. 4, the melting temperature is measured as 1265 K. The T_g/T_m value is determined to be 0.67 which is high enough to be judged as a high glass-forming alloy. In addition, it is seen that the melting of the $\text{Ni}_{53}\text{Nb}_{20}\text{Ti}_{10}\text{Zr}_8\text{Co}_6\text{Cu}_3$ alloy occurs through a single stage, indicating that the Ni-based alloy lies in the vicinity of eutectic point.

Figure 5 shows tensile and compressive stress-elongation curves at room temperature for the cast bulk glassy $\text{Ni}_{53}\text{Nb}_{20}\text{Ti}_{10}\text{Zr}_8\text{Co}_6\text{Cu}_3$ alloys. The tensile test was made for the sheet sample with a gauge dimension of 5 mm in width, 10 mm in length and 1 mm in thickness, while the compressive test used the rod sample with a gauge dimension of 2 mm in diameter and 4 mm in length. The Young's modulus (E), tensile fracture strength ($\sigma_{t,f}$) and tensile fracture elongation including elastic elongation ($\varepsilon_{t,f}$) are 140 GPa, 2700 MPa and 2.1%, respectively. Similarly, the compressive fracture strength ($\sigma_{c,f}$) and compressive fracture elongation including elastic elongation ($\varepsilon_{c,f}$) are 3010 MPa and 2.4%, respectively. The tensile fracture strength of 2700 MPa is believed to be the highest for all bulk glassy alloys reported up to date. Figure 6 summarizes the mechanical properties in the relation between tensile fracture strength ($\sigma_{t,f}$) and Young's modulus (E), together with the previous data for other typical bulk glassy al-

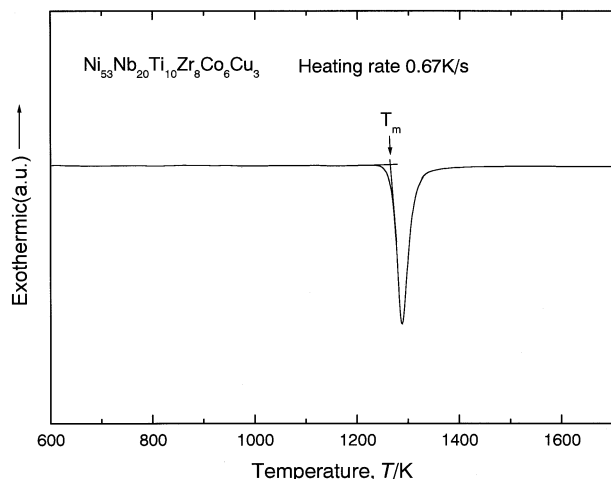


Fig. 4 DTA curve of the $\text{Ni}_{53}\text{Nb}_{20}\text{Ti}_{10}\text{Zr}_8\text{Co}_6\text{Cu}_3$ alloy.

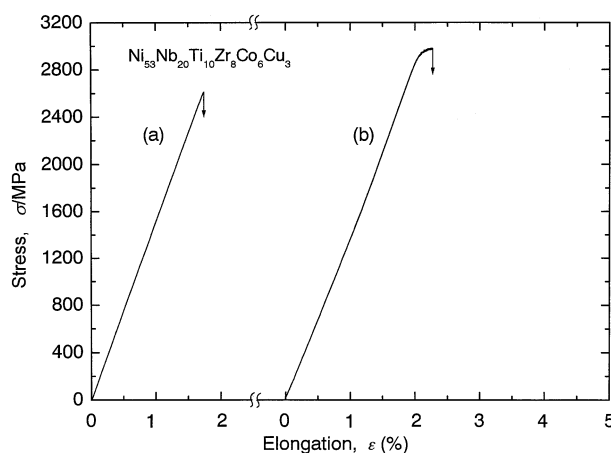


Fig. 5 Stress-elongation curves of the cast bulk glassy $\text{Ni}_{53}\text{Nb}_{20}\text{Ti}_{10}\text{Zr}_8\text{Co}_6\text{Cu}_3$ alloys in rod and sheet shapes. (a) Tensile deformation of the sheet with a thickness of 1 mm, and (b) compressive deformation of the rod with a diameter of 2 mm.

loys and conventional crystalline alloys.²⁴⁾ It is seen that the present data lie in the previous linear relation, indicating that the mechanism for the high strength of the present bulk glassy alloy may be the same as that for the other bulk glassy alloys.

With the aim of investigating the origin for the high strengthening mechanism for the present bulk glassy alloys, the tensile fracture strength was plotted as a function of T_g or T_m in Fig. 7. Although some scatterings are seen, one can see distinct linear relations between $\sigma_{t,f}$ and T_g or T_m . Considering that T_g or T_m is dominated by the bonding forces among the constituent elements of the glassy alloys,³⁰⁾ the high $\sigma_{t,f}$ value of the present Ni–Nb–Ti–Zr–Co–Cu bulk glassy alloys is concluded to reflect the strong bonding forces among the main four elements of Ni, Nb, Ti and Zr. This is consistent with the data that the negative heats of mixing for their main atomic pairs are 30 kJ/mol for Ni–Nb, 49 kJ/mol for Ni–Zr and 28 kJ/mol for Ni–Ti.³¹⁾ On the contrary, the good ductility leading to appreciable plastic elongation seems to reflect the nearly zero heat of mixing for Nb–Zr, Zr–Ti and Nb–Ti atomic pairs.³¹⁾ In addition, one can recognize that the constituent elements have significant atomic size ratios of 1.18 for Nb/Ni, 1.28 for Zr/Ni, 1.18 for Ti/Ni, 1.09 for Zr/Nb and 1.09 for Zr/Ti.³²⁾ Based on these fundamental data on

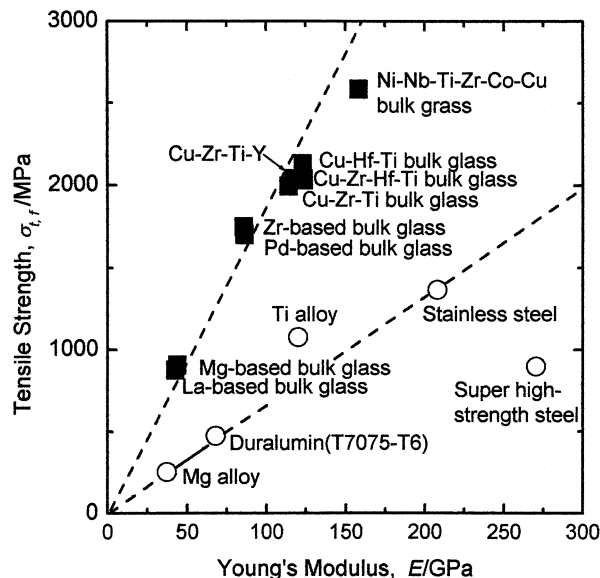


Fig. 6 Relation between tensile fracture strength ($\sigma_{t,f}$) and Young's modulus (E) for the cast bulk glassy $\text{Ni}_{53}\text{Nb}_{20}\text{Ti}_{10}\text{Zr}_8\text{Co}_6\text{Cu}_3$ alloy sheet. The data of the other bulk glassy alloys and conventional crystalline alloys are also shown for comparison.

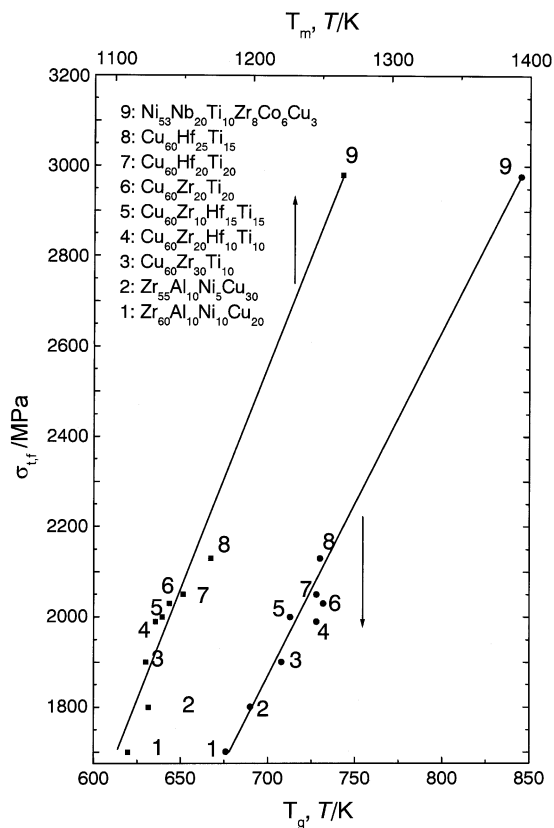


Fig. 7 Relation between tensile fracture strength ($\sigma_{t,f}$) and glass transition temperature (T_g) or melting temperature (T_m) for the cast bulk glassy $\text{Ni}_{53}\text{Nb}_{20}\text{Ti}_{10}\text{Zr}_8\text{Co}_6\text{Cu}_3$ alloy sheet. The data of the other bulk glassy alloys are also shown for comparison.

the bonding nature and atomic size ratios, the high strength combined with good ductility is due to the unique alloy components leading to the mixed state of the attractive and repulsive bonding natures for the constituent elements with significantly different atomic size ratios. Finally, it is important to point out that the present Ni–Nb–Ti–Zr base alloy

satisfies almost the following three criteria for achievement of high glass-forming ability,^{21–24)} i.e., (1) multi-component alloy system consisting of more than three elements, (2) significant atomic size ratios above about 12% among the main constituent elements, and (3) negative heats of mixing among their main elements. It is further said that the addition of the special elements with a nearly zero heat of mixing and significantly different atomic size is important for the achievement of high strength and good ductility for Ni-based metal-metal type alloys. The subsequent search for a new alloy in the framework of this concept is expected to prepare bulk Fe- and Co-based glassy alloys with high strength and good ductility.

4. Conclusions

With the aim of developing a new bulk glassy alloy with high tensile strength and good ductility in ferrous alloy group, we examined glass-forming ability and mechanical properties for Ni-based alloys where the three empirical criteria for formation of bulk glassy alloys are almost satisfied. The results obtained are summarized as follows.

(1) Bulk glassy alloys were formed in Ni–Nb–Ti–Zr base system by the copper mold casting and mold-clamp casting methods. The maximum diameter was 3 mm for glassy $\text{Ni}_{53}\text{Nb}_{20}\text{Ti}_{10}\text{Zr}_8\text{Co}_6\text{Cu}_3$ alloy.

(2) The bulk glassy alloy exhibits the distinct glass transition, followed by a large supercooled liquid region and then crystallization. The glass transition temperature (T_g) and the supercooled liquid region defined by the difference between T_g and crystallization temperature (T_x), $\Delta T_x (= T_x - T_g)$ are 846 and 51 K, respectively. In addition to the large ΔT_x value, the bulk glassy alloy is located in the vicinity of eutectic composition and exhibits high reduced glass transition temperature (T_g/T_m) of 0.67.

(3) The Ni-based bulk glassy alloy exhibits Young's modulus of 140 GPa, tensile fracture strength of 2700 MPa, tensile fracture elongation of 2.1%, compressive fracture strength of 3010 MPa and compressive fracture elongation of 2.4%. The high tensile fracture strength is roughly proportional to Young's modulus, glass transition temperature and melting temperature which reflect the bonding forces among the constituent elements.

(4) The finding of the new Ni-based glassy alloys with high glass-forming ability and good mechanical properties is encouraging for future development of bulk glassy alloys

which can be used as high-strength structural materials.

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